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## Crystal Structure

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# 7-Vinyl-8-aza-7-deaza-2'-deoxyadenosine monohydrate 

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In the title compound, 4-amino-1-(2-deoxy- $\beta$-D-eythro-pentofuranosyl)-3-vinyl-1 H -pyrazolo[3,4- $d$ ]pyrimidine monohydrate, $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$, the conformation of the glycosyl bond is anti. The furanose moiety is in an $S$ conformation with an unsymmetrical twist, and the conformation at the exocyclic $\mathrm{C}-\mathrm{C}(\mathrm{OH})$ bond is $+s c$ (gauche, gauche). The vinyl side chain is bent out of the heterocyclic ring plane by $147.5(5)^{\circ}$. The three-dimensional packing is stabilized by $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}, \mathrm{O}-$ $\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds.

## Comment

The 7 -substituted 8 -aza-7-deazapurine $2^{\prime}$-deoxyribonucleosides (3-substituted pyrazolo[3,4- $d$ ]pyrimidine $2^{\prime}$-deoxyribonucleosides) are studied as analogues of natural DNA constituents. Compound (II) (Seela \& Steker, 1985; Seela, Zulauf et al., 1999; Seela \& Kaiser, 1988) has attracted attention as it is an ideal shape mimic of $2^{\prime}$-deoxyadenosine. Purine numbering is used throughout the manuscript. The 7 -substituted derivatives of compound (II) have a stabilizing effect on oligonucleotide duplexes (Seela, Becher \& Zulauf, 1999; Seela \& Zulauf, 1999). Thus, the 7-position of an 8-aza-7-deazapurine $2^{\prime}$-deoxyribonucleoside is an attractive site for modification as it is the 5 -position of a pyrimidine nucleoside (Gourlain et al., 2001). Cheng et al. (1976) reported that 5-vinyl-2'-deoxyuridine, (III), has the capacity for viral and tumour inhibition. A single-crystal X-ray analysis of (III) was reported by Hamor et al. (1978). The present manuscript reports the single-crystal X-ray structure of the title compound, (I), containing a vinyl side chain in the 7 -position.

Canonical purine $2^{\prime}$-deoxyribonucleosides tend to adopt an anti conformation. The orientation of the base relative to the sugar moiety (syn/anti) of purine nucleosides is defined by the torsion angle $\chi\left(\mathrm{O}^{\prime}-\mathrm{C}^{\prime}-\mathrm{N} 9-\mathrm{C} 4\right)$ (IUPAC-IUB Joint Commission on Biochemical Nomenclature, 1983). In the crystal structure of (I) (Fig. 1 and Table 1), the conformation of the glycosyl bond is between anti and high anti $[\chi=$
$-106.9(5)^{\circ}$ ]. A similar conformation was observed for the parent 8 -aza-7-deazapurine $2^{\prime}$-deoxyribonucleoside (II) $[\chi=$ -106.3 (2) ${ }^{\circ}$; Seela, Zulauf et al., 1999]. Halogen substituents at the 7-position drive the conformation to high anti $[\chi=$ -74.1 (4) (7-bromo) and -73.2 (4) (7-iodo) (Seela et al., 2000)]. This phenomenon might be caused by stereoelectronic effects of the base. As the vinyl group has an electron-withdrawing influence on the heterocyclic base, the $\mathrm{p} K_{a}$ value of protonation is decreased from 4.0 in (II) to 3.54 in (I). The glycosyl bond length in (I) $\left[\mathrm{N} 9-\mathrm{C}^{\prime}=1.460(4) \AA\right]$ is slightly longer than that in (II) $[1.442$ (2) $\AA$ ].

(I)
Purine numbering

(II)
Systematic numbering

The pseudorotation phase, $P$, and the puckering amplitude, $\tau$, angles (Rao et al., 1981) show that the sugar ring of (I) adopts an $S$ conformation, with an unsymmetrical twist of the $\mathrm{C} 4^{\prime}$-endo- $\mathrm{C}^{\prime}$-exo bond (between ${ }^{3} E$ and ${ }_{3} T^{4}$ ), having a $P$ value of $205.6(4)^{\circ}$ and a $\tau$ value of 30.2 (2) $)^{\circ}$. In (II), the sugar ring conformation is ${ }_{3} T^{2}\left[P=182.2\right.$ (2) ${ }^{\circ}$ and $\left.\tau=41.2(2)^{\circ}\right]$. The conformation about the $\mathrm{C}^{\prime}-\mathrm{C}^{\prime}$ bond of (I) is $+s c$ (gauche, gauche), with a dihedral angle, $\gamma$, of 42.1 (4) ${ }^{\circ}$, whereas in (II), the $\mathrm{C} 4^{\prime}-\mathrm{C} 5^{\prime}$ bond adopts a $-a p$ (trans) conformation, with $\gamma$ equal to $-178.73(16)^{\circ}$.

The base moiety of (I) is nearly planar, but the vinyl side chain deviates from the plane. The r.m.s. deviation of ring atoms $\mathrm{N} 9, \mathrm{~N} 8, \mathrm{C} 7, \mathrm{C} 5, \mathrm{C} 6, \mathrm{~N} 1, \mathrm{C} 2, \mathrm{~N} 3$ and C 4 from the leastsquares plane through these atoms is $0.012 \AA$, with a maximum deviation of -0.024 (4) $\AA$ (atom C6). Atom $\mathrm{C}^{\prime}$ is displaced from this plane by -0.014 (6) $\AA$. The C5-C7$\mathrm{C} 71=\mathrm{C} 72$ torsion angle linking the vinyl group and the heterocycle is 147.5 (5) ${ }^{\circ}$. In (III), the vinyl group is inclined by $12^{\circ}$ to the pyrimidine ring plane (Hamor et al., 1978). Such a deviation was also observed in the propynyl group in 8-aza-7-deaza-7-propynyladenosine (Lin et al., 2005). The conformation of the conjugated diene system $(\mathrm{N} 8=\mathrm{C} 7-\mathrm{C} 71=\mathrm{C} 72)$ is $s-Z$. This conformation occurs because of the steric repulsion between the vinyl chain and the 6 -amine group.

In the three-dimensional network, the bases are stacked (3.3 $\AA$ apart; Fig. 2). Each nucleoside is connected to one water molecule, which is coordinated by hydrogen bonds. The water molecule acts as an acceptor (O10) of a hydrogen bond from atom $\mathrm{O} 3^{\prime}$ and as a donor of two H atoms in bonds to atoms N3 and O5' (Table 2). Hence, three nucleosides are connected by one water molecule. There is an intramolecular

## Figure 1

A perspective view of (I). Displacement ellipsoids of non-H atoms are drawn at the $50 \%$ probability level and H atoms are shown as spheres of arbitrary size.


Figure 2
The crystal packing of (I), viewed down the $a$ axis, showing the hydrogen bonds as dashed lines.
hydrogen bond between the $\mathrm{O5}^{\prime}-\mathrm{H}^{\prime}$ group and atom N 8 , which we have not observed in related nucleosides. Other inter- and intramolecular hydrogen bonds are summarized in Table 2.

## Experimental

Compound (I) was synthesized according to a known procedure (Seela \& Zulauf, 1998) and was crystallized as the monohydrate from aqueous methanol. For the diffraction experiment, a single crystal was fixed at the top of a Lindemann capillary with epoxy resin.

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$
$M_{r}=295.31$
Orthorhombic, $P 2_{1} 2_{1} 2_{1}$
$a=6.6513$ (8) A
$b=9.603$ (5) $\AA$
$c=21.481$ (3) A
$V=1372.1$ (7) $\AA^{3}$
$Z=4$
$D_{x}=1.430 \mathrm{Mg} \mathrm{m}^{-3}$
Data collection
Bruker $P 4$ diffractometer
$2 \theta / \omega$ scans
2800 measured reflections
2100 independent reflections
1180 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.040$
$\theta_{\text {max }}=29.0^{\circ}$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.058$
$w R\left(F^{2}\right)=0.113$
$S=1.02$
2100 reflections
205 parameters
H atoms treated by a mixture of independent and constrained refinement

$$
\begin{aligned}
& h=-1 \rightarrow 9 \\
& k=-1 \rightarrow 13 \\
& l=-29 \rightarrow 1 \\
& 3 \text { standard reflections } \\
& \quad \text { every } 9 \text { reflections } \\
& \text { intensity decay: none }
\end{aligned}
$$

Mo $K \alpha$ radiation
Cell parameters from 28 reflections
$\theta=4.9-12.5^{\circ}$
$\mu=0.11 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, colourless
$0.42 \times 0.28 \times 0.20 \mathrm{~mm}$

$$
\begin{gathered}
w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0349 P)^{2}\right] \\
\text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
(\Delta / \sigma)_{\max }<0.001 \\
\Delta \rho_{\max }=0.22 \mathrm{e} \AA^{-3} \\
\Delta \rho_{\min }=-0.21 \mathrm{e}^{-3}
\end{gathered}
$$

Extinction correction: SHELXL97
Extinction coefficient: 0.0074 (11)

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right.$ ).

| C7-C71 | 1.461 (4) | N8-N9 | 1.370 (3) |
| :---: | :---: | :---: | :---: |
| C71-C72 | 1.306 (5) | $\mathrm{N} 9-\mathrm{Cl}^{\prime}$ | 1.460 (4) |
| N8-C7-C5 | 109.3 (3) | C5-C7-C71 | 129.6 (3) |
| N8-C7-C71 | 121.1 (3) | C72-C71-C7 | 123.8 (4) |
| N8-C7-C71-C72 | -30.6 (7) | $\mathrm{C} 4-\mathrm{N} 9-\mathrm{Cl}^{\prime}-\mathrm{C}^{\prime}$ | 134.0 (5) |
| C5-C7-C71-C72 | 147.5 (5) | $\mathrm{N} 8-\mathrm{N} 9-\mathrm{C1}^{\prime}-\mathrm{C}^{\prime}{ }^{\prime}$ | -48.3 (6) |
| $\mathrm{C} 4-\mathrm{N} 9-\mathrm{Cl}^{\prime}-\mathrm{O} 4{ }^{\prime}$ | -106.9 (5) | $\mathrm{O} 4^{\prime}-\mathrm{C4}^{\prime}-\mathrm{C} 5^{\prime}-\mathrm{O}^{\prime}$ | -76.7 (4) |
| $\mathrm{N} 8-\mathrm{N} 9-\mathrm{Cl}^{\prime}-\mathrm{O}^{\prime}$ | 70.7 (5) | $\mathrm{C3}^{\prime}-\mathrm{C4}^{\prime}-\mathrm{C5}^{\prime}-\mathrm{O}^{\prime}$ | 42.1 (4) |

In the absence of suitable anomalous scattering, Friedel equivalents could not be used to determine the absolute structure. Therefore, Friedel pairs were merged before the final refinements. The known configuration of the parent molecule was used to define the enantiomer of the final nucleoside. All H atoms were initially found in a difference Fourier synthesis. H atoms bonded to C and N atoms were placed in idealized positions $(\mathrm{C}-\mathrm{H}=0.93-0.98 \AA$ and $\mathrm{N}-\mathrm{H}=$ $0.86 \AA$ ) and constrained to ride on their parent atoms, with $U_{\text {iso }}(\mathrm{H})$ values of $1.2 U_{\text {eq }}(\mathrm{C})$ and $1.5 U_{\text {eq }}(\mathrm{N})$. The H -atom positions were found

Table 2
Hydrogen-bonding geometry ( $\AA{ }^{\circ},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N} 6-\mathrm{H} 64 \cdots \mathrm{O} 4^{\text {i }}$ | 0.86 | 2.39 | 2.995 (4) | 128 |
| $\mathrm{N} 6-\mathrm{H} 6 \mathrm{~B} \cdots \mathrm{O}^{\text {/ii }}$ | 0.86 | 2.24 | 3.010 (4) | 150 |
| $\mathrm{O}^{\prime}-\mathrm{H}^{\prime} \cdots$. O 10 | 0.82 (2) | 1.88 (2) | 2.698 (5) | 174 (5) |
| O5'-H5'. ${ }^{\prime}$ N8 | 0.83 (4) | 2.02 (4) | 2.832 (4) | 164 (4) |
| $\mathrm{O} 10-\mathrm{H} 10 A \cdots \mathrm{~N} 3^{\text {iii }}$ | 0.91 (3) | 1.93 (3) | 2.831 (4) | 169 (4) |
| $\mathrm{O} 10-\mathrm{H} 10 \mathrm{~B} \cdots \mathrm{O}^{\text {iv }}$ | 0.91 (3) | 1.86 (3) | 2.755 (4) | 169 (5) |

Symmetry codes: (i) $x-\frac{1}{2}, \frac{3}{2}-y, 1-z$; (ii) $\frac{3}{2}-x, 2-y, \frac{1}{2}+z$; (iii) $1-x, \frac{1}{2}+y, \frac{1}{2}-z$; (iv) $1-x, y-\frac{1}{2}, \frac{1}{2}-z$.
in a difference Fourier map and were geometrically idealized and constrained (DFIX). The $U_{\text {iso }}(\mathrm{H})$ values were constrained to be 1.5 times $U_{\text {eq }}(\mathrm{O})$.

Data collection: XSCANS (Siemens, 1996); cell refinement: XSCANS; data reduction: SHELXTL (Sheldrick, 1997); program(s) used to solve structure: $S H E L X T L$; program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 1999).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: GA1089). Services for accessing these data are described at the back of the journal.

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